23. Photochemical Reactions

141st Communication¹)

Photochemistry of α,β -Unsaturated δ,ε -Epoxyketones of the Ionone Series

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On n,π^* - as well as on π,π^* -excitation, the 4,5-epoxy- α -ionones (E)-1, (E)-2, and (E)-3 undergo (E)/(Z)isomerization and subsequent γ -H-abstraction leading to the corresponding 4-hydroxy- β -ionones (E/Z)-9, (E/ Z)-13, and (E/Z)-17 as primary photoproducts. On photolysis of (E)-3, as an additional primary photoproduct, the β,γ -unsaturated δ,ϵ -epoxy ketone 18 was obtained. The other isolated compounds, namely the 2H-pyrans 10A + B and 14A + B as well as the *retro*- γ -ionones 11 and 15A + B, represent known types of products, which are derived from the 4-hydroxy- β -ionones (E/Z)-9 and (E/Z)-13, respectively.

1. Introduction. – It has been shown previously [2] that irradiation of β , γ -epoxy-ketones such as I (Scheme 1) initially causes Norrish-type-I bond cleavage leading to a



diradical **a** which undergoes subsequent opening of the oxirane ring. The hereby formed acyl alkoxy diradical **b** reacts further to give various types of products. These findings made it of interest to examine, whether an oxirane cleavage could also be induced by initial abstraction of a γ -H-atom in α,β -unsaturated δ,ϵ -epoxyketones, and what type of products would be formed. We decided to investigate the behavior of the 4,5-epoxy-4,5-dihydro- α -ionones (E)-1, (E)-2 and (E)-3 (Scheme 2), since α -ionone ((E)-4) undergoes photoisomerization involving γ -H-abstraction [3]. In addition, comparison of the photochemical behavior of (E)-1 with that of its methyl homologs (E)-2 and (E)-3 was expected to give information about the influence of a CH₃-substituent in α - or β -position of the enone chromophore. This comparison was of particular inter-

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est, since, in contrast to α -ionone ((E)-4), 8-methyl- α -ionone ((E)-5)⁴) did not show photo-isomerization via γ -H-abstraction [5]. Furthermore, from NMR measurements at low temperature it was evident that, in contrast to (E)-4 and (E)-5, in 7-methyl- α ionone ((E)-6), the rotation of the side chain around the C(6),C(7)-bond is hindered due to steric interaction of the CH₃-group at C(7) with the geminal CH₃-groups [6].

2. Preparation and Stereochemical Assignment of the Epoxyenones (E)-1, (E)-2 and (E)-3. – Compound (E)-1 was obtained by treatment of (E)-4 with monoperphthalic acid in Et₂O according to Karrer and Stürzinger [7]. Analogously, the previously unknown epoxyenone (E)-2 was prepared in 82% yield by stereoselective epoxidation of (E)-5. The reaction of (E)-6 [8] with monoperphthalic acid led to a 1:1 mixture of the diastereomeric epoxyenones (E)-3 (36%) and (E)-7 (36%; see Scheme 2). The epoxidation of (Z)-6 was fairly stereoselective, however; thus the epoxides (Z)-3 and (Z)-7 were obtained in 76% and 10% yields, respectively. Treatment of both (E)- and (Z)-3 with a ca. 0.3M solution of NaOMe in MeOH at reflux temperature led to a 9:2 mixture of (E/Z)-3 (ca. 88%). On the other hand, analogous treatment of their diastereomers (E)- and (Z)-7 afforded in each case the bicyclic enol ether 8 in ca. 95% yield. This different behavior of the diastereometric epoxyenones (E/Z)-3 and (E/Z)-7 allows their stereochemistry to be assigned. Only (Z)-7 fulfils the stereoelectronic requirements for facile oxirane ring opening, *i.e.*, the *trans*-diaxial arrangement of the two O-atoms involved. Therefore, the epoxide O-atom and the enone side chain must be in trans-relation in (E/Z)-7, and in cis-relation in (E/Z)-3. Furthermore, the epoxyenones (E/Z)-3 were obtained in 23% yield by alkylation of the epoxyenone (E)-1, furnishing further confirmation of the previously assigned stereochemistry of the latter [7].

3. Photolyses. – Irradiation of ca. 0.05M solutions of the epoxyenones (E)-1, (E)-2, and (E)-3 in MeCN and $\lambda > 280$ nm (n, π^* -excitation) and $\lambda = 254$ nm (π,π^* -excitation), respectively, up to ca. 90% conversion (details are given in the *Exper. Part*) gave the following product distributions.

⁴) In ionone derivatives, numbering according to the carotinoid nomenclature is used [4].

3.1. Photolyses of (E)-1 (Scheme 3). - 3.1.1. n,π^* -Excitation: (Z)-1 (1%), (E)-9 (13%), 10A (33%), 10B (37%), 11 (3%), 12A (1%), and 12B (3%)⁵).

3.1.2. π,π^* -Excitation: (Z)-1 (6%), (E)-9 (14%), 10A (4%), 10B (5%), and 11 (33%)⁵).

3.2. Photolyses of (E)-2 (see Scheme 3). 3.2.1. n,π^* -Excitation: (Z)-2 (2%), (E)-13 (26%), (Z)-13⁶) (15%), 14A (9%), 14B (11%), 15A (9%), 15B (9%), 16A (4%), and 16B (2%).



3.2.2. π,π^* -Excitation: (Z)-2 (6%), (E)-13 (17%), (Z)-13 (30%)⁶), 14A (2%), 14B (2%), 15A (14%), and 15B (16%).

3.3. Photolyses of (E)-3 (see Scheme 4). - 3.3.1. n,π^* -Excitation: (Z)-3 (39%), (E)-17 (16%), (Z)-17A (13%), (Z)-17B (25%), 18 (1%), and additional photoproducts arising from (E/Z)-17 (2%)⁷).



3.3.2. π,π^* -Excitation: (Z)-3 (19%), (E)-17 (12%), (Z)-17A (12%), (Z)-17B (22%), 18 (8%), and additional photoproducts arising from (E/Z)-17 (4%)⁷).

4. Structure of the Products. – The structures of all new compounds were deduced from their spectral data. Since OH-free analogs of most of the products obtained here are known from the photolyses of β -ionone [10], only the most relevant spectral data are discussed herein together with the chemical transformations which confirmed the assigned structures. Full spectral data and the assignment of the NMR data are given in the *Exper. Part.*

⁷) The results of the photolyses of (E/Z)-17 will be reported in a forthcoming paper [9].

⁵) With the exception of (Z)-1, the same products were obtained on photolysis of (E)-9 under these conditions (see *Exper. Part*).

⁶) Chromatography fractions containing (Z)-13 gradually changed to a mixture 14A/14B in a 3:1 equilibrium ratio.

Epoxyenones (E/Z)-1, (E/Z)-2, (E/Z)-3, and (E/Z)-7 (Scheme 2 and 3). The stereochemical relation of the epoxide moiety and the enone side chain of compounds (E/Z)-3 and (E/Z)-7 was established by their reactivity upon treatment with base and on the transformation of the epoxyenone (E)-1 of known configuration [7] to (E/Z)-3 (see Sect. 2). The inspection of models indicated that the epoxidation of (E)-5 occurs from the same side as in (E)-4, thus leading to the epoxyenone (E)-2 with the same relative configuration as (E)-1. The configuration of the enone double bond of (E)- and (Z)-1 was assigned unequivocally from the ¹H-NMR-coupling constants of the olefinic H-atoms ((E)-1: J = 16 [7b], (Z)-1: J = 11.5). On the other hand, the configuration of the double bond of (E/Z)-2, (E/Z)-3, and (E/Z)-7 was correlated with the chemical shift of the allylic H-C(6)⁴). Due to the anisotropic effect of the carbonyl group, this signal of (Z)-2 is shifted downfield to 2.85 ppm whereas the corresponding signal of (E)-2 appears at 2.47 ppm. This effect is even more profound in the case of (Z)-3 and (Z)-7, in the spectra of which the signals of H-C(6)⁴) are shifted in both cases to *ca.* 4.2 ppm, whereas the corresponding signals of (E)-7 appear at 2.15 and 2.35 ppm, respectively.

4-Hydroxy- β -ionones (E)-17 and (Z)-17A + B (Scheme 4) were independently prepared via alkylation of (E)-9 [11]. It is noteworthy that two conformers (Z)-17A and (Z)-17B arising from hindered rotation around the C(6),C(7)-bond⁴) could be separated by chromatography, however, heating them separately to 80° for 5 h led in both cases to 1:1 mixtures.



2H-Pyrans 10A + B and 14A + B (Scheme 3). The structures were deduced from their spectral data and by comparing them with that of 19 (Scheme 5), a known product of the photolysis of β -ionone [10]. Reaction of 10A + B and 14A + B with ¹O₂ gave the endoperoxides 20A + B and 21A + B, respectively. Analogous to the acid-catalyzed transformation of 22 \rightarrow 23 [12], the endoperoxides 20A + B were converted to the acyclic triketone 24. Photolysis ($\lambda > 280$ nm) of 10A and 10B led to a 1:2 mixture of the tricyclic oxetans 12A + B. The acetate 25B derived from the alcohol 12B was hydrogenated to the saturated compound 26, which was also obtained as a photoproduct of 4-acetoxy-7,8-dihydro- β -ionone (27) [13].

5. Discussion. $-n,\pi^*$ - as well as π,π^* -excitation of the epoxyenone (E)-1, (E)-2, and (E)-3 causes rapid (E)/(Z)-isomerization and formation of the 4-hydroxy- β ionones (E/Z)-9, (E/Z)-13, and (E/Z)-17 as primary photoproducts. On photolysis of (E)-3, as an additional primary photoproduct, the β,γ '-unsaturated δ,ε -epoxy ketone 18 (Scheme 4) was obtained in low yield. The formation of these compounds can be explained by γ -H-abstraction of the epoxyenones (Z)-1, (Z)-2, and (Z)-3 at C(6) leading to the diradical intermediates of type c. Analogously to the oxirane cleavage $\mathbf{a} \rightarrow \mathbf{b}$ (Scheme 1), the intermediate c undergoes scission of the C(5),O-bond of the oxirane, which is followed by H-transfer (Scheme 6). Compound 18 (Scheme 4) arises by an alternative γ -H-abstraction from CH₃-C(7)-group in (E)-3.





The other isolated compounds 10A + B, 11, 12A + B, 14A + B, 15A + B, and 16A + B are all secondary photoproducts derived from (E/Z)-9, (E/Z)-13, and (E/Z)-17, respectively.

Comparison of the behavior of the epoxyenones (E)-1, (E)-2, and (E)-3 with that of the parent α -ionones (E)-4, (E)-5, and (E)-6 (Scheme 7) is of interest. Thus, it has



been reported that (E)-4 efficiently undergoes photo-isomerization to the *retro-\alpha*-ionones (E/Z)-28 via a γ -H-abstraction at C(6) [3]. On the other hand, photolysis of (E)-5 in neutral solvents led to the compounds 29 and 30 involving as first step a 1,2-H-shift ((E)-5 \rightarrow d; see Scheme 7)⁸). The *retro-\alpha*-ionones (E/Z)-31 were only formed on photolysis of (E)-5 in acidic or basic media. Finally, (E)-6 was shown to be quite unreactive. Besides (E)/(Z)-isomerization, only the formation of the deconjugated compound 32 (Scheme 7) arising from γ -H-abstraction at CH₃--C(7) was observed to a small extent [11]. The lack of formation of the *retro-\alpha*-ionones (E/Z)-33 is apparently due to an unfavorable steric interaction between the two substituents at C(7) and the ring CH₃-groups.

On the basis of the differing behavior of (E)-4, (E)-5, and (E)-6, it was surprising that irradiation of the derived epoxyenones (E)-1, (E)-2, and (E)-3 should lead to the same types of primary products. However, the formation of (E/Z)-9, (E/Z)-13, and (E/Z)-17 may not necessarily imply the same reaction mechanism. In particular, for the transformation of (E)-2 \rightarrow (E/Z)-13 an alternative mechanism could be considered (Scheme 6). Thus, analogous to the transformation of (E)-5 \rightarrow d (Scheme 7), a 1,2-H-shift in (E)-2 could lead to the diradical intermediate e. Instead of ring closure analogous to $d \rightarrow 29 + 30$ (Scheme 7) leading to compounds 34 and 35, the intermediate e could undergo rapid cleavage of the C(5),O-bond of the oxirane followed by a 1,5-H-shift furnishing (E)- or (Z)-13°).

The 4-hydroxy- β -ionones (E/Z)-9, (E/Z)-13 and (E/Z)-17 show the same behavior as their analogs without an OH-function in 4-position (cf. [5] [9] [10]). Thus, the 2Hpyrans 10A + B and 14A + B are formed by thermal or photochemical cyclization of the dienones (Z)-9 and (Z)-13, respectively. On the other hand, cyclization of the dienones (Z)-17A + B to the 2H-pyrans 36A + B does not occur due to an unfavorable steric interaction between the CH₃-group at C(7) and the geminal CH₃-groups. Due to this steric repulsion, the two conformers (Z)-17A and (Z)-17B arising from hindered rotation around the C(6),C(7)-bond could be isolated.

Also in analogy to the β -ionone derivatives without an OH-function in 4-position, the hydroxylated compounds (*E*)-9, (*E*)-13, and (*E*)-17 undergo a 1,5-sigmatropic H-shift leading to the *retro-y*-ionones 11, 15A + B, and 37⁷)¹⁰).

Furthermore, on n,π^* -excitation of the epoxyenones (E)-1 and (E)-2, the compounds 12A + B and 16A + B, respectively, are formed in low yield (Scheme 3). The tricyclic oxetanes 12A + B are products of an electrocyclic reaction of the 2*H*-pyrans

¹⁰) Etoh et al. [14] reported that π,π^* -excitation of the hydroxyacetate (E)-38 gave rise to a photochemically induced dehydration via a y-H-abstraction furnishing the compound (E)-39 and its secondary products 40A + B and 41 (Scheme 8).



⁸) The labile compound **30** which was previously postulated [5] has been recently isolated by flash chromatography [11].

⁹) This hypothesis may be supported by the detection of traces (ca. 2%) of two diastereomeric compounds of structure **35**, they could, however, not be isolated in pure form.

10A + B as was demonstrated by photolysis of the latter. The compounds 16A + B presumably arise from the homoconjugated ketones 15A + B via a 1,3-acyl shift, which was previously reported for retro- α -ionone (E/Z)-28 [3].

Conclusion. – The aforementioned results demonstrate that, on n,π^* - as well as on π,π^* -excitation, the 4,5-epoxy- α -ionones (E/Z)-1, (E/Z)-2, and (E/Z)-3 may be converted efficiently to the corresponding 4-hydroxy- β -ionones (E/Z)-9, (E/Z)-13, and (E/Z)-14 via initial γ -H-abstraction.

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Experimental Part

General. See [15], except as noted below. Analytical gas chromatography was performed using a 25 m \times 0.33 mm Ucon 50 HB 5100 glass capillary. Column chromatography was carried out on silica gel 60 Merck 0.040–0.063 mm, 230–400 mesh ASTM (SiO₂) according to [16] ('flash chromatography'). Analytically pure samples were obtained, in general, after repeated column chromatography on SiO₂; in some cases further purification was necessary with an HPLC (*Du Pont Instruments*, Model 830, UV detector), using a 25 cm \times 23.6 mm SiO₂ column. In general, ¹H-NMR spectra were taken on a *Bruker WP-80 CW* (80 MHz) or *WM 300* (300 MHz) instrument in CDCl₃-soln. or, exceptionally (as indicated below), in CCl₄-soln. on a *Varian HA-100* instrument (100 MHz).

Alcohols were acetylated by reaction with 2 equiv. of Ac_2O in *ca*. 0.5M pyridine soln. In some cases a catalytic amount of *N*,*N*-dimethylaminopyridine was added. The mixture was worked up in Et₂O and the org. phase washed with sat. aq. NaHCO₃ and sat. aq. CuSO₄.

1. Preparations. – 1.1. Epoxidation of (E)-3-Methyl-4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-3-buten-2one(= (E)-8-Methyl- α -ionone, (E)-5). To a soln. of (E)-5 (513 mg, 2.49 mmol; containing ca. 7% of the isomeric ethyl ketone 42)¹¹) in Et₂O (20 ml) a soln. of monoperphthalic acid in Et₂O [17] (0.50M, 6.00 ml, 3.00 mmol) was added at 0°, the mixture first stirred for 1 h at 0° and then for 12 h at r.t. The org. phase was washed with Na₂S₂O₃-soln. (sat. aq.), NaHCO₃-soln. (sat.) and worked up. Chromatography (Et₂O/pentane 4:1) afforded a mixture of (E)-2 and 43 (453 mg, 82%).

Treatment of crude (*E*)-**2** (1.26 g, 5.67 mmol) with NaOMe (380 mg, 7.03 mmol) in MeOH (25 ml) at reflux temp. for 24 h afforded after workup and chromatography (SiO₂, Et₂O/pentane 1:1) pure (*E*)-**2** (1.09 g, $87\%)^{11}$).

(E,1' RS,2' RS,3' SR)-4-(2',3'-Epoxy-2',6',6'-trimethylcyclohexyl)-3-methyl-3-buten-2-one ((E)-2). B.p. 80°/ 0.01 Torr. UV (0.249 mg in 25 ml MeCN): 232 (15600). UV (6.010 mg in 10 ml MeCN): 317 (50). IR: 3055w, 2960s, 2925m, 2865w, 1670s, 1640w, 1470w (sh), 1460w (sh), 1445m, 1435w, 1420w (sh), 1385w, 1375m, 1365m, 1345w, 1315w (sh), 1305w (sh), 1300w, 1290w (sh), 1260w (sh), 1250m, 1240w (sh), 1205w, 1180w (sh), 1175w,

¹¹) Compounds (*E*)-5 and 42 as well as the epoxides (*E*)-2 and 43 could not be separated. However, on treatment of (*E*)-2 and 43 with NaOMe only 43 was transformed into the hydroxydienone 44 as described for the conversion of α -iononepoxide (*E*)-1 to (*E*)-4-hydroxy- β -ionone (*E*)-9 [7]. Compound (*E*)-2 could easily be separated from 44 by chromatography.



1145w, 1090w, 1050w, 980w, 965w, 910m, 880w. ¹H-NMR: 0.74, 0.93, 1.21 (3s, 2 CH₃-C(6'), CH₃-C(2')); 0.9–1.6 (m, 2H-C(5')); 1.82 (d, J = 1.5, CH₃-C(3)); 1.8–2.1 (m, 2H-C(4')); 2.32 (s, 3H-C(1)); 2.47 (d, J = 10.0, H-C(1')); 3.06 (dd, $J_1 = J_2 = 2.0$, H-C(3')); 6.62 (dq, $J_1 = 10.0$, $J_2 = 1.5$, H-C(4)). ¹³C-NMR (20 MHz): 12.0, 24.3, 25.6, 26.6, 28.5 (5q, 5 CH₃); 21.9, 29.2 (2t, C(4'), C(5')); 32.0 (s, C(6')); 47.8 (d, C(1')); 59.3 (s, C(2')); 59.7 (d, C(3')); 139.0 (s, C(3)); 141.7 (d, C(4)); 199.6 (s, C(2)). MS: 222 (2, M^+ , C₁₄H₂₂O₂), 207 (3), 193 (6), 179 (10), 137 (16), 125 (20), 123 (60), 111 (18), 109 (36), 95 (14), 81 (15), 69 (16), 55 (19), 43 (100), 41 (15). Anal. calc. for C₁₄H₂₂O₂ (222.23): C 75.63, H 9.97; found: C 75.94, H 10.01.

1.2. Epoxidation of (E)-4-(2,6,6-Trimethyl-2-cyclohexen-1-yl)-3-penten-2-one (=(E)-7-Methyl- α -ionone; (E)-6). A soln. of (E)-6 (1.120 g, 5.43 mmol) in Et₂O (50 ml) was treated with a soln. of monoperphthalic acid in Et₂O (0.50 m, 13.0 ml, 6.50 mmol) and worked up as described for (E)-5 (see Sect. 1.1). Chromatography $(Et_2O/pentane 4:1)$ of the crude product afforded (E)-3 (432 mg, 36%) and (E)-7 (438 mg, 36%).

(E,1' RS,2' RS,3' SR)-4-(2',3'-Epoxy-2',6',6'-trimethylcyclohexyl)-3-penten-2-one ((E)-3). B.p. 70°/0.01 Torr. UV (0.339 mg in 25 ml MeCN): 241 (12800). UV (3.644 mg in 2 ml MeCN): 325 (65). IR: 2970s, 2950s (sh), 2930s, 2870m, 2850m (sh), 1710w (sh), 1685s, 1605s, 1470w (sh), 1460m (sh), 1450m, 1435m, 1425m (sh), 1385m, 1375m, 1365m, 1320w, 1305w, 1290w, 1205s, 1180m, 1165m, 1105w, 1065w, 1015w, 1005w, 965m, 950w, 900w, 865w. ¹H-NMR: 0.92 (s, 2 CH₃-C(2')); 0.80–1.70 (m, 2H-C(5')); 1.80–2.20 (m, 2H-C(4')); 2.15 (s, H-C(1')); 2.23 (s, 3H-C(1)); 2.28 (d, J = 1.5, 3H-C(5)); 2.94 (dd, J₁ = J₂ = 3, H-C(3')); 6.29 (m, w₂ = 3, H-C(3)). ¹³C-NMR (20 MHz): 21.1, 23.5, 24.2, 31.5, 32.1 (5q, 5 CH₃); 21.6, 34.2 (zt, C(4'), C(5')); 33.0 (s, C(6')); 57.9, 58.8 (2d, C(1'), C(3')); 59.1 (s, C(2')); 128.5 (d, C(3)); 157.1 (s, C(4)); 198.4 (s, C(2)). MS: 222 (2, M⁺, C₁₄H₂₂O₂), 179 (10), 151 (12), 137 (12), 125 (17), 124 (10), 123 (70), 109 (47), 107 (12), 95 (11), 93 (11), 81 (15), 69 (17), 55 (20), 43 (100), 41 (29). Anal. calc. for C₁₄H₂₂O₂ (222.33): C 75.63, H 9.97; found: C 75.61, H 9.89.

(E,1' RS,2' SR,3' RS)-4-(2',3'-Epoxy-2',6',6'-trimethylcyclohexyl)-3-penten-2-one ((E)-7). B.p. 70°/0.01 Torr. UV (0.380 mg in 25 ml MeCN): 241 (13400). UV (3.118 mg in 2 ml MeCN): 326 (65). IR: 3000w (sh), 2960s, 2930s, 2870m, 1685s, 1600s, 1475w, 1460m (sh), 1445m (sh), 1435m, 1420m (sh), 1385s, 1380s, 1365m, 1350m, 1310w, 1230w (sh), 1200s, 1170m, 1160m, 1150m (sh), 1070w, 1015w br., 960m, 900w, 875w, 855w (sh), 845w. ¹H-NMR: 0.82, 1.00 (2s, 2 CH₃-C(6')); 1.23 (s, CH₃-C(2)); 1.10–1.50 (m, 2H–C(5')); 1.90–2.30 (m, 2H–C(4')); 2.16 (m, $w_{1/2}$ = 6, 3H–C(5)); 2.21 (s, 3H–C(1)); 2.35 (s, H–C(1')); 3.05 (dd, J_1 = J_2 = 2.5, H–C(3')); 6.18 (m, $w_{1/2}$ = 6, H–C(3)). ¹³C-NMR (75 MHz, CD₃CN): 20.2, 24.1, 24.1, 29.3, 31.0 (5q, 5 CH₃)); 22.3, 32.8 (2t, C(4'), C(5')); 32.5 (s, C(6')); 59.6 (s, C(2')); 59.7, 60.6 (2d, C(1'), C(3')); 129.8 (d, C(3)); 157.1 (s, C(4)); 198.5 (s, C(2)). MS: 222 (3, M^+ , C₁₄H₂₂O₂, 207 (4), 189 (5), 179 (10), 161 (15), 151 (11), 137 (10), 125 (12), 123 (53), 111 (13), 109 (26), 107 (11), 95 (11), 93 (16), 55 (20), 43 (100), 41 (30). Anal. calc. for C₁₄H₂₂O₂ (22.2.33): C 75.63, H 9.97; found: C 75.47, H 9.92.

1.3. Epoxidation of (Z)-6. A soln. of (Z)-6 (1.782 g, 8.64 mmol) in Et₂O (50 ml) was treated with a soln. of monoperphthalic acid in Et₂O (0.50M, 20.7 ml, 10.4 mmol) and worked up as described for (E)-5 (see Sect. 1.1). Chromatography (Et₂O/pentane 9:1 to 4:1) of the crude product yielded (Z)-3 (1.466 g, 76%) and (Z)-7 (196 mg, 10%).

(Z,I' RS,Z' RS,J' SR)-4-(Z',J'-Epoxy-2',6',6'-trimethylcyclohexyl)-3-penten-2-one ((Z)-3). B.p. 80°/0.01 Torr. UV (0.303 mg in 25 ml MeCN): 244 (10900). UV (5.061 mg in 2 ml MeCN): 327 (55), end absorption to 390. IR: 3000w (sh), 2960s, 2940s (sh), 2920s, 2870m, 2850w (sh), 1680s, 1600s, 1470w (sh), 1460m (sh), 1450m, 1435m, 1420m, 1385m, 1375m, 1360m, 1350m, 1300w, 1230w, 1220w, 1200m, 1190m, 1170s, 1140w, 1095w, 1060m, 1015w, 970m, 920w, 910w, 900m, 855m. ¹H-NMR: 0.83, 0.94 (2s, 2 CH₃-C(6')); 1.18 (s, CH₃-C(2')); 0.80–1.80 (m, 2H-C(5')); 1.80–2.10 (m, 2H-C(4')); 2.04 (d, J = 1.5, 3H-C(5)); 2.17 (s, 3H-C(1)); 2.94 (dd, $J_1 = J_2 = 2.5$, H-C(3')); 4.22 (s, H-C(1')); 6.32 (m, $w_4 = 3$, H-C(3)). ¹³C-NMR (20 MHz): 23.7, 24.1, 25.8, 30.2 (5q, 2q overlapping, 5 CH₃); 22.0, 32.7 (2t, C(4'), C(5')); 32.2 (s, C(6')); 45.9 (d, C(1')); 58.0 (d, (C(3')); 58.8 (s, C(2')); 128.9 (d, C(3)); 158.1 (s, C(4)); 198.3 (s, C(2)). MS: 222 (12, M^+ , C₁₄H₂₂O₂), 207 (20), 204 (31), 188 (20), 165 (10), 161 (31), 151 (14), 149 (31), 148 (91), 147 (19), 137 (17), 133 (50), 125 (11), 124 (12), 123 (88), 121 (12), 119 (18), 109 (29), 108 (26), 107 (16), 105 (33), 95 (12), 91 (26), 77 (21), 55 (22), 43 (100), 41 (41). Anal. cale. for C₁₄H₂₂O₂ (222.33): C 75.63, H 9.97; found: C 75.65, H 9.87.

(Z, I' RS, 2' SR, 3' RS)-4-(2', 3'-Epoxy-2', 6', 6'-trimethylcyclohexyl)-3-penten-2-one ((Z)-7). UV (0.340 mg in 25 ml MeCN): 243 (9900). UV (4.888 mg in 2 ml MeCN): 327 (50), end absorption to 390. IR: 3000w, 2960s, 2930s, 2910m (sh), 2870m, 1685s, 1605s, 1470w, 1460m (sh), 1450m, 1440m, 1420m, 1385m, 1380m, 1370m, 1360m, 1350m, 1310w, 1240w, 1215w, 1205w, 1180s, 1170s, 1100w, 1070w, 965m, 870w. ¹H-NMR: 0.81, 1.13 (2s, 2 CH₃-C(6')); 1.19 (s, CH₃-C(2')); 0.80-1.80 (m, 2H-C(5')); 1.92 (d, J = 1.5, 3H-C(5)); 2.21 (s, 3H-C(1)); 1.80-2.20 (m, 2H-C(4')); 3.01 (m, $w_{Y_2} = 7$, H-C(3')); 4.23 (s, H-C(1')); 6.23 (m, $w_{Y_2} = 4$, H-C(3)). MS: 222

(12, M^+ , $C_{14}H_{22}O_2$), 204 (6), 189 (5), 161 (8), 18 (17), 139 (20), 133 (11), 124 (12), *123* (100), 109 (10), 105 (9), 43 (46), 41 (15). Anal. calc. for $C_{14}H_{22}O_2$ (222.33): C 75.63, H 9.97; found: C 75.48, H 9.90.

1.4. Treatment of (E)- and (Z)-3 with Base. a) A suspension of (Z)-3 (3.00 g, 13.5 mmol), NaOMe (3.00 g, 55.5 mmol) and MeOH (150 ml) was heated under reflux for 5 h. The mixture was concentrated *in vacuo* and the residue extracted with Et_2O /pentane 1:1. Workup and chromatography (Et_2O /pentane 1:9) afforded (E)-3 (2.17 g, 72%) and (Z)-3 (0.47 g, 16%). b) The analogous reaction with (E)-3 (723 mg, 3.25 mmol) lead to a mixture of (E)- and (Z)-3 (662 mg, 92%, 9:2).

1.5. Treatment of (E)- and (Z)-7 with Base. a) A suspension of (E)-7 (763 mg, 3.43 mmol), NaOMe (770 mg, 14.3 mmol) and MeOH (50 ml) was heated under reflux for 3 h. The mixture was concentrated *in vacuo* and the residue extracted with Et_2O /pentane 1:1. Workup afforded compound 8 (750 mg, 98%). b) The reaction of (Z)-7 (220 mg, 0.99 mmol) with NaOMe (212 mg, 3.92 mmol) in MeOH (20 ml) was complete after refluxing for 30 min, affording 8 (206 mg, 94%).

10-Hydroxy-1,3,7,7-tetramethyl-5-methylidene-2-oxabicyclo[4.4.0]dec-3-ene (8). B.p. 80°/0.01 Torr. UV (0.514 mg in 25 ml MeCN): 257 (13400). UV (3.630 mg in 2 ml MeCN): end absorption to 300. IR: 3635m, 3500w (br.), 3080w, 3050w, 2980s, 2970s, 2940s, 2920s, 2870s, 1745w, 1650s, 1605w, 1470w, 1455m, 1450m, 1440m, 1430m, 1415m, 1385s (sh), 1380s, 1370m, 1360m, 1340s, 1290m, 1240w, 1200m, 1170s, 1120s, 1065s, 990s, 960m, 950m, 930w, 910w, 885m, 870s, 855m. ¹H-NMR: 0.80, 0.92 (2s, 2 CH₃-C(7)); 1.20 (s, CH₃-C(1)); 1.72 (s, CH₃-C(3)); 0.80-2.50 (m, 2H-C(8), 2H-C(9), OH); 1.95 (m, $w_{V_4} = 3$, H-C(6)); 3.75 (dd, $J_1 = J_2 = 2.5$, H-C(10)); 4.46, 4.82 (2m, $w_{V_2} = 4$, 2 CH₂=C(5)); 5.22 (m, $w_{V_2} = 3$, H-C(4)). ¹³C-NMR (20 MHz): 20.1, 21.8, 22.6 (3q, 3 CH₃); 25.3, 33.4 (2t, C(8), C(9)); 32.1 (q, CH₃-C(3)); 33.1 (s, C(7)); 49.3 (d, C(6)); 72.3 (d, C(10)); 78.5 (s, C(1)); 103.3 (d, C(4)); 109.1 (t, C=C(5)); 138.5 (s, C(5)); 150.2 (s, C(3)). MS: 222 (12, M^+ , C₁₄H₂₂O₂, 139 (20), 124 (12), 123 (100), 43 (38). Anal. calc. for C₁₄H₂₂O₂ (222.33): C 75.63, H 9.97; found: C 75.34, H 10.04.

1.6. Preparation of (E/Z)-3 from (E)-4-(2,3-Epoxy-2,6,6-trimethylcyclohexyl)-3-buten-2-one (=(E)-4,5-Epoxy- α -ionone; (E)-1). To a suspension of CuI (5.08 g, 26.7 mmol) in Et₂O (250 ml), a solution of MeLi in Et₂O (1.6M, 32.7 ml, 52.3 mmol) was added at -40° and the mixture was stirred for 20 min. (E)-1 [7] (4.54 g, 21.8 mmol) in Et₂O (30 ml) was then added slowly at -40°, and the mixture was allowed to warm to -10° over 20 min. Again at -40° a soln. of phenylselenyl bromide in THF (45 ml) (prepared by reaction of diphenyl diselenide (8.41 g, 26.9 mmol) and Br₂ (0.95 ml, 18.5 mmol) in THF at 0°) was added rapidly. The cooling bath was removed and the mixture was allowed to warm up to r.t., poured into Et₂O/pentane 1:1 (300 ml) and aq. NH₄Cl (sat., 200 ml) and worked up. The crude product was dissolved in pyridine (6.40 ml) and CH₂Cl₂ (300 ml), and H₂O₂ (15%, 32 ml, 141 mmol) was added dropwise at r.t. (the reaction was highly exothermic). After 1 h at r.t., aq. NaHCO₃ (10%, 70 ml) was added, the org. phase washed with 1M aq. HCl and worked up. Column chromatography (acetone/CH₂Cl₂ 1:100) yielded the starting material ((E)-1, 160 mg, 4%), (E)-3 (292 mg, 6%), (Z)-3 (805 mg, 17%), 45¹²) (1.64 g, 34%), and 46¹²) (164 mg, 3%).

2. Photolysis Experiments. – 2.1. Photolyses of (E)-1 in MeCN. – 2.1.1. At $\lambda > 280$ nm. A solution of (E)-1 [7] (5.00 g, 24.0 mmol) in MeCN (600 ml) was irradiated (Pyrex, lamp B, 97% conversion). Chromatography (Et₂O/pentane 1:1), gave mixed fractions; from their ¹H-NMR analysis, the following product distribution was determined: (Z)-1 (1%), 10A (33%), 10B (37%), 11 (3%), (E)-9 [18] (13%), 12A (1%) and 12B (3%). Compounds 11 and 12B were purified and isolated as their acetates.

(Z, I' RS, 2' RS, 3' SR)-4-(2', 3' - Epoxy-2', 6', 6' - trimethylcyclohexyl)-3-buten-2-one ((Z)-1). B.p. 70°/0.01 Torr. UV (0.254 mg in 25 ml pentane): 231 (9400). UV (5.678 mg in 10 ml pentane): 335 (40). IR: 3020w, 2960s, 2930s, 2870m, 2850m (sh), 1715w (sh), 1690s, 1685m (sh), 1665m 1610m, 1460m (sh), 1445m, 1435m (sh), 1410m, 1385w (sh), 1380m, 1365m, 1350m, 1305w, 1245w, 1170s, 1145w, 1095w, 965m, 895m. ¹H-NMR (C₆D₆): 0.80, 0.82 (2s, 2 CH₃-C(6')); 1.24 (s, CH₃-C(2')); 1.78 (s, 3H-C(1)); 0.75 (dddd, $J_1 = 13.4$, $J_2 = 5.7$, $J_3 = 3.6$, $J_4 = 1.2$, H-C(5')); 1.37 (ddd, $J_1 = 13.4$, $J_2 = 10.8$, $J_3 = 5.5$, H-C(5')); 1.55 (dddd, $J_1 = 15.3$, $J_2 = 10.8$, $J_3 = 5.7$, $J_4 = 2.5$, H-C(4')); 1.77 (dddd, $J_1 = 15.3$, $J_2 = 5.5$, $J_3 = 3.6$, $J_4 = 1.5$, H-C(4')); 2.74 (m, $w_{1/2} = 5.5$, H-C(3')); 3.90 (dm, J = 10.8, $w_{1/2} = 2.5$, H-C(1')); 6.01 (*AB*-system, J = 11.5, $\delta_A = 5.93$, H-C(3), $\delta_B = 6.09$, B part split

¹²) The spectral data of these compounds are given in the Ph. D. thesis of P.M.



to d, J = 10.8, H–C(4)). ¹³C-NMR (C₆D₁₂): 24.0, 26.8, 28.1, 31.4 (4q, 4 CH₃); 22.6, 29.3 (2t, C(4'), C(5')); 31.5 (s, C(6')); 44.8 (d, C(1')); 58.8 (s, C(2')); 59.3 (d, C(3')); 128.7 (d, C(4)); 145.8 (d, C(3)); 196.7 (s, C(2)). MS: 208 (19, M^+ , C₁₃H₂₀O₂), 193 (18), 190 (12), 179 (28), 175 (23), 165 (16), 151 (13), 147 (16), 134 (33), 123 (18), 109 (73), 95 (24), 91 (16), 86 (18), 84 (35), 79 (13), 69 (16), 57 (82), 56 (64), 55 (25), 43 (100), 42 (31), 41 (73). Anal. calc. for C₁₃H₂₀O₂ (208.30): C 74.96, H 9.68; found: C 74.79, H 9.73.

10-Hydroxy-1,3,7,7-tetramethyl-2-oxabicyclo[4.4.0]deca-3,5-diene, Isomer A (10A). IR: 3680s, 3070w, 3050w, 2960s, 2935s, 2920s (sh), 2870s, 1730w, 1660s, 1650m (sh), 1600m, 1465m (sh), 1460m (sh), 1450s, 1445m (sh), 1430m, 1385s, 1375s, 1360s, 1340m, 1330m, 1315s, 1280m, 1255s, 1235m, 1205s, 1180m, 1155s, 1145m, 1125m, 1075s, 1060s, 1050s (sh), 1030m, 1000m (sh), 995m, 975m, 965s, 945m, 930w, 915m, 885w, 875w, 860w, 845m, 670m, 620w. ¹H-NMR (C₆D₆, 300 MHz): 0.98, 1.12 (2s, 2 CH₃-C(7)); 1.22 (s, CH₃-C(1)); 1.59 (m, $w_{1/2} = 2.5$, CH₃-C(3)); 1.01 (td, overlapping with s, $J_1 = 13$, $J_2 = 7$, H-C(8)); 1.58 (m, H-C(8)); 1.91 (dq, $J_1 = 14$, $J_2 = 4$, H-C(9)); 2.17 (dt, broadened to m, $J_1 = 14$, $J_2 = 4$, $w_{1/2} = 3$, H-C(9)); 2.90 (m, $w_{1/2} = 20$, OH); 3.76 (m, $w_{1/2} = 8$, H-C(10)); 4.93 (dq, $J_1 = 6.0$, $J_2 = 1.1$, H-C(4)); 5.71 (dm, J = 6, $w_{1/2} = 2$, H-C(5)). ¹³C-NMR (C₆D₁): 19.8, 22.6, 30.7, 31.6 (4q, 4 CH₃); 25.0, 32.8 (2t, C(8), C(9)); 34.6 (s, C(7)); 72.5 (d, C(10)); 81.1 (s, C(1)); 99.0, 116.3 (2d, C(4), C(5)); 137.9, 148.7 (2s, C(3), C(6)).

Isomer B (10B). IR: 3610*m*, 3070*w*, 3050*w*, 2980*s* (sh), 2960*s*, 2940*s*, 2920*s* (sh), 2870*m*, 1720*w*, 1660*m*, 1650*m* (sh), 1600*w*, 1460*m*, 1445*m*, 1395*w*, 1380*m*, 1370*m*, 1350*m*, 1320*m*, 1315*m*, 1285*m*, 1260*m*, 1210*w*, 1180*w*, 1150*m* (sh), 1125*s*, 1085*m*, 1060*s*, 1025*m*, 1000*m*, 975*w*, 940*w*, 935*w*, 865*w*. ¹H-NMR (C₆D₆, 300 MHz): 0.94, 0.98 (2*s*, 2 CH₃-C(7)); 1.48 (*s*, CH₃-C(1)); 1.68 (*s*, CH₃-C(3)); 1.10-1.32 (*m*, 2H-C(8)); 1.51-1.80 (*m*, overlapping with *s*, 2H-C(9)); 2.05-2.35 (*m*, OH); 3.99 (*dd*, $J_1 = 12.0$, $J_2 = 5.0$, H--C(10)); 4.93 (*dq*, $J_1 = 6.0$, $J_2 = 1.1$, H-C(4)); 5.60 (*dm*, J = 6.0, $w_{V_2} = 2.0$, H--C(5)). ¹³C-NMR (C₆D₁₂): 17.2, 19.9, 30.7, 31.5 (*4q*, 4 CH₃); 26.9, 38.1 (2*t*, C(8), C(9)); 35.3 (*s*, C(7)); 77.6 (*d*, C(10)); 83.0 (*s*, C(1)); 98.5, 116.3 (2*d*, C(4), C(5)); 138.3, 149.8 (2*s*, C(3), C(6)). MS: 208 (6, M^+ , $C_{13}H_{20}O_2$), 193 (7), 190 (13), 175 (8), 165 (3), 151 (5), 147 (6), 135 (13), *134* (100), 133 (11), 119 (7), 106 (19), 91 (24), 57 (13), 43 (17), 41 (16). Anal. calc. for C₁₃H₂₀O₂ (208.30): C 74.96, H 9.68; found: C 74.86, H 9.85.

4,4-Dimethyl-2-methylidene-3-(3'-oxobutylidene)cyclohexyl Acetate (Acetate of 11). B.p. 70°/0.01 Torr. UV (0.290 mg in 25 ml MeCN): 220 (sh, 6000). UV: (2.653 mg in 2 ml MeCN): 284 (290). IR: 3080w (sh), 2960m, 2945m, 2920m (sh), 2870w, 2850w, 1735s, 1720s, 1635w, 1470w, 1460w, 1405w, 1385w, 1370m, 1360m, 1310w, 1325s, 1160m, 1070w (sh), 1040m, 1030m, 1015m (sh), 990w, 965w, 910w (br.), 875w. ¹H-NMR: 1.03 (s, 2 CH₃-C(4)); 1.2-2.2 (m, 2H-C(5), 2H-C(6)); 2.01, 2.08 (2s, 3H-C(4'), CH₃CO₂); 3.20 (d, J = 7, 2H-C(2')); 4.72, 5.18 (2m, $w_{1/4} = 4$, 2H-C=C(2)); 5.03-5.23 (m, overlapping with mat 5.18, H-C(1)); 5.48 (t, J = 7, H-C(1')). ¹³C-NMR: 21.1, 27.0, 27.0, 29.6 (4q, 4 CH₃); 28.5, 37.1, 43.5 (3t, C(2'), C(5), C(6)); 37.7 (s, C(4)); 7.5.0 (d, C(1)); 112.7 (t, C=C(2)); 115.0 (d, C(1')); 143.2, 149.1 (2s, C(2), C(3)); 169.9 (s, CH₃CO₂); 206.9 (s, C(3')). MS: 250 (2, M^+ , $C_{15}H_{22}O_3$), 235 (4), 208 (3), 190 (18), 175 (11), 165 (11), 148 (22), 147 (62), 134 (17), 133 (63), 119 (23), 105 (42), 92 (31), 91 (29), 55 (14), 43 (100), 41 (17). Anal. calc. for C₁₅H₂₂O₃ (250.34): C 71.97, H 8.86; found: C 72.09, H 8.88.

7-Hydroxy-4,6,10,10-tetramethyl-5-oxatricyclo[4.4.0.0^{1,4}]dec-2-ene, Isomer A (12A). B. p. 90°/0.01 Torr. IR: 3560w, 3510w (br.), 3120w, 3040w, 2960s, 2940s, 2920s (sh), 2870s, 1460w (sh), 1450m, 1435w, 1390m, 1370s, 1365m, 1315m, 1305w, 1280m, 1240w, 1135w, 1090w, 1065s, 1060s (sh), 1040m, 1000w, 970w, 960w, 900w, 885m, 870m, 840m. ¹H-NMR: 0.84, 0.89 (2s, 2 CH₃-C(1)); 1.18 (s, CH₃-C(6)); 1.49 (s, CH₃-C(4)); 1.0-2.2 (m, 2H-C(8), 2H-C(9)); 3.25 (m, $w_{1/2} = 6$, OH); 3.64 (m, $w_{1/2} = 10$ with fine structure, H-C(7)); 6.46 (AB, J = 3, $\delta_A = 6.43$, $\delta_B = 6.49$, H-C(2), H-C(3)). ¹³C-NMR: 17.7, 22.6, 26.7, 26.7 (4q, 4 CH₃); 25.1, 31.9 (2t, C(8), C(9)); 30.6 (s, C(10)); 63.9 (s, C(1)); 71.1 (d, C(7)); 79.6, 90.6 (2s, C(4), C(6)); 140.6, 143.5 (2d, C(2), C(3)). MS: 208 (11, M^+ , C₁₃H₂₀O₂), 193 (46), 151 (23), 147 (14), 139 (17), 137 (17), 135 (22), 134 (18), 133 (13), 123 (21), 121 (24), 119 (22), 109 (57), 107 (22), 105 (23), 95 (18), 93 (19), 91 (33), 81 (16), 79 (20), 77 (19), 55 (19), 53 (14), 43 (100), 41 (32). Anal. calc. for C₁₃H₂₀O₂ (208.30): C 74.96, H 9.68; found: C 74.89, H 9.56.

4,6,10,10-Tetramethyl-5-oxatricyclo[4.4.0.0^{1,4}]dec-2-en-7-yl Acetate, Isomer B (**25B**). B.p. 90°/0.01 Torr. IR: 3040w, 2960m, 2940m (sh), 2925m, 2870w, 2860w (sh), 1740s, 1455w, 1445w (sh), 1370m, 1365m (sh), 1315w, 1295w, 1240s, 1190w, 1160w, 1135w, 1125w (sh), 1090m, 1065w, 1045s, 1020m, 1000m, 990w (sh), 975w, 915w, 890m, 870w, 840w. ¹H-NMR (300 MHz): 0.88, 0.98 (2s, 2 CH₃-C(10)); 1.22, 1.52 (2s, CH₃-C(4), CH₃-C(6)); 1.10-1.85 (m, H-C(8), 2 H-C(9)); 1.90-2.04 (m, H-C(8)); 2.11 (s, CH₃CO₂); 5.05 (dd, $J_1 = 9.5$, $J_2 = 4.0$, H-C(7)); 6.50 (AB, J = 3.0, $\delta_A = 6.48$, $\delta_B = 6.52$, H-C(2), H-C(3)). ¹³C-NMR: 17.9, 21.3, 21.6, 27.4, 28.4 (5q, 5 CH₃); 23.0, 32.9 (2t, C(8), C(9)); 30.3 (s, C(10)); 64.8 (s, C(1)); 74.0 (d, C(7)); 78.7, 88.7 (2s, C(4), C(6)); 139.7, 143.7 (2d, C(2), C(3)); 170.9 (s, CH₃CO₂-). MS: 250 (2, M^+ , $C_{15}H_{22}O_3$), 235 (21), 207 (13), 190 (47), 175 (40), 151 (11), 147 (23), 135 (25), 134 (47), 109 (26), 105 (17), 91 (30), 43 (100), 41 (21). Anal. calc. for $C_{15}H_{22}O_3$ (250.34): C 71.97, H 8.86; found: C 71.98, H 8.75. 2.1.2. At $\lambda = 254$ nm. A soln. of (E)-1 (1.00 g, 4.80 mmol) in MeCN (200 ml) was irradiated (quartz, lamp A, 91% conversion). Chromatography of the mixture (Et₂O/pentane 1:1) gave mixed fractions from which the following product distribution was determined (¹H-NMR): (Z)-1 (6%), 10A (4%), 10B (5%), 11 (33%) and (E)-9 [18] (14%).

2.2. Photolysis of (E)-1 in Acetone at $\lambda > 280$ nm. A soln. of (E)-1 (1.00 g, 4.80 mmol) in acetone (100 ml) was irradiated (*Pyrex*, lamp *B*, 2 h). Chromatography (Et₂O/pentane 1:2) afforded (*E*)-1 (184 mg, 18%), and (*Z*)-1 (736 mg, 74%).

2.3. Photolyses of (E)-9 in MeCN. – 2.3.1. At $\lambda > 280$ nm. A soln. of (E)-9 (1.12 g, 5.38 mmol) in MeCN (100 ml) was irradiated (*Pyrex*, lamp B, 84% conversion). After chromatography (Et₂O/pentane 1:1 to 3:1), the following product distribution was determined (¹H-NMR): **10A** (17%), **10B** (22%), **11** (31%), **12A** (3%), and **12B** (8%).

2.3.2. At $\lambda = 254$ nm. A soln. of (E)-9 (412 mg, 1.98 mmol) in MeCN (200 ml) was irradiated (quartz, lamp A, 88% conversion). Chromatography (Et₂O/pentane 1:1 to 3:1) afforded **10A** (3%), **10B** (3%) and **11** (57%).

2.4. Photolyses of 10A and 10B in MeCN at $\lambda > 280$ nm. Two separate samples of 10A and 10B (50 mg, 0.24 mmol) in MeCN (10 ml) were irradiated (Pyrex, lamp B, ca. 50% conversion, ca. 40°). Capillary GC and ¹H-NMR analysis showed for both samples the same ratio (1:2) of the oxetanes 12A and 12B as well as a ratio (ca. 1:1) of the pyrans 10A and 10B.

2.5. Photolyses of (E)-2 in MeCN. – 2.5.1. At $\lambda > 280$ nm. A soln. of (E)-2 (5.00 g, 22.5 mmol) in MeCN (720 ml) was irradiated (*Pyrex*, lamp B, 99% conversion). After chromatography (Et₂O/pentane 1:1 to 3:1), the following product distribution was determined (¹H-NMR): (Z)-2 (2%), 14A (9%), 14B (11%), (E)-13 (26%), (Z)-13 (15%), 15A (9%), 15B (9%), 16A (4%), 16B (2%), and 35 (ca. 2%).

(Z,I' RS,2' RS,3' SR)-4-(2',3'-Epoxy-2',6',6'-trimethylcyclohexyl)-3-methyl-3-buten-2-one ((Z)-2). B.p. 65°/0.01 Torr. UV (0.439 mg in 25 ml MeCN): 237 (5400). UV (4.791 mg in 2 ml MeCN): end absorption to 390. IR: 2960s, 2925s, 2870m, 1690s, 1685s (sh), 1620m (br.), 1460m, (sh), 1450m, 1435m (sh), 1375m, 1365s, 1350m, 1305w, 1225m, 1200m, 1180m, 1145w, 1125m, 1090w, 1065w, 965m, 910m, 875w. ¹H-NMR: 0.73, 0.86 (2s, 2 CH₃-C(6')); 1.20 (s, CH₃-C(2')); 0.7-1.6 (m, 2H-C(5')); 1.8-2.1 (m, 2H-C(4')); 1.99 (d, J = 1.5, CH₃-C(3)); 1.22 (s, 3H-C(1)); 2.85 (d, J = 11.0, H-C(1')); 3.01 (dd, J₁ = J₂ = 2, H-C(3')); 5.61 (dq, J₁ = 11.0, J₂ = 1.5, H-C(4)). ¹³C-NMR (20 MHz): 21.2, 24.2, 26.9, 28.0, 29.8 (5q, 5 CH₃); 22.0, 28.7 (2t, C(4'), C(5')); 31.4 (s, C(6')); 45.7 (d, C(1')); 59.5 (s, C(2')); 59.8 (d, C(3')); 135.1 (d, C(4)); 137.6 (s, C(3)); 203.0 (s, C(2)). MS: 222 (6, M^+ , C₁₄H₂₂O₂, 207 (18), 153 (11), 137 (26), 125 (14), 123 (85), 111 (14), 109 (35), 107 (13), 95 (14), 69 (15), 55 (23), 53 (12), 43 (100), 41 (33). Anal. calc. for C₁₄H₂₂O₂ (222.33): C 75.63, H 9.97; found: C 75.43, H 9.93.

10-Hydroxy-1,3,4,7,7-pentamethyl-2-oxabicyclo[4.4.0]deca-3,5-diene, Isomer A (14A). IR: 3575s, 3030w (sh), 2960s, 2935s, 2920s, 2870s, 1670m, 1600w, 1460m (sh), 1450m, 1435m (sh), 1390s, 1385s (sh), 1365m, 1295m (sh), 1280s, 1225s, 1210m (sh), 1190m, 1150s, 1120m, 1080m (sh), 1060s, 1000m, 970m, 915w, 890w, 860m. ¹H-NMR (80 MHz): 1.11, 1.16, 1.27 (3s, CH₃-C(1), 2 CH₃-C(7)); 0.9–2.3 (m, 2H–C(8), 2H–C(9)); 1.68, 1.79 (2m, $w_{1/2}$ = 3, CH₃-C(3), CH₃-C(4)); 2.95 (m, $w_{1/2}$ = 3, OH); 3.75 (m, $w_{1/2}$ = 7, H–C(10)); 5.64 (s, H–C(5)). ¹³C-NMR (20 MHz): 15.4, 16.2, 22.6, 30.3, 31.2, (5q, 5 CH₃); 24.4, 32.4 (2t, C(8), C(9)); 34.1 (s, C(7)); 72.3 (d, C(10)); 79.3 (s, C(1)); 104.6 (s, C(4)); 120.5 (d, C(5)); 138.5, 142.2 (2s, C(3), C(6)). MS: 222 (38, M^+ , C₁₄H₂₂O₂), 208 (15), 207 (99), 177 (18), 165 (27), 163 (50), 153 (23), 151 (23), 150 (38), 149 (31), 148 (26), 137 (26), 135 (20), 123 (80), 122 (22), 109 (19), 105 (19), 91 (20), 77 (17), 55 (16), 43 (100), 41 (26).

Isomer B (14B). IR: 3610*m*, 3040*w* (sh), 2980*s* (sh), 2960*s*, 2940*s*, 2920*s*, 2860*s*, 1740*w*, 1670*m*, 1600*w*, 1460*m*, 1440*m* (sh), 1380*m*, 1365*m*, 1320*w*, 1275*m*, 1255*m*, 1230*m*, 1190*m*, 1150*m*, 1130*s*, 1070*s*, 1060*s*, 1050*s*, 1030*m*, 960*w*, 865*m*. ¹H-NMR: 1.06, 1.10 (2*s*, 2 CH₃-C(7)); 1.26 (*s*, CH₃-C(1)); 1.00–2.10 (*m*, 2H–C(8), 2H–C(9)); 1.64, 1.74 (2*m*, $w_{\frac{1}{2}}$ = 3, CH₃-C(3), CH₃-C(4)); 2.87 (*m*, $w_{\frac{1}{2}}$ = 9, OH); 3.78–4.00 (*m*, H–C(10)); 5.52 (*s*, H–C(5)). ¹³C-NMR (20 MHz): 15.3, 16.3, 17.0, 30.3, 31.2 (5*q*, 5 CH₃); 26.2, 37.4 (2*t*, C(8), C(9)); 34.7 (*s*, C(7)); 77.6 (*d*, C(10)); 81.5 (*s*, C(1)); 104.2 (*s*, C(4)); 120.3 (*d*, C(5)); 139.3, 143.1 (2*s*, C(3), C(6)). MS: 222 (35, M^+ , C₁₄H₂₂O₂), 208 (16), 207 (100), 189 (9), 177 (13), 165 (28), 163 (24), 153 (23), 151 (22), 149 (30), 148 (26), 137 (27), 123 (80), 109 (18), 105 (19), 91 (19), 77 (16), 55 (15), 43 (90), 41 (27).

(E)-4-(3'-Hydroxy-2', 6', 6'-trimethyl-1'-cyclohexenyl)-3-methyl-3-buten-2-one ((E)-13). B.p. 100°/0.01 Torr. UV (0.517 mg in 25 ml MeCN): 228 (8400), 266 (3800). UV (2.130 mg in 2 ml MeCN): end absorption to 380. IR: 3610w, 3480w (br.), 2960s, 2930s, 2860m, 1670s, 1620w, 1445m (sh), 1435m, 1380m, 1360s, 1335w, 1285w, 1245s, 1200w, 1170w, 1100m, 1070m, 1030m, 1020m, 990m, 960m, 905w, 865w. ¹H-NMR: 0.96, 1.01 (2s, 2 CH₃-C(6')); 1.60, 1.65 (2m, $w_{Y_2} = 3$, CH₃-C(3), CH₃-C(2')); 1.20-2.10 (m, 2H-C(4'), 2H-C(5')); 2.32 (s, 3H-C(1)); 2.25-2.45 (m, OH); 4.00 (m, $w_{Y_2} = 12$, H-C(3')); 7.04 (m, $w_{Y_2} = 6$, H-C(4)). ¹³C-NMR (contaminated with 30% of **15A**): 12.2, 17.3, 25.0, 26.6, 27.6 (5q, 5CH₃); 27.9, 33.9 (2t, C(4'), C(5')); 34.6 (s, C(6')); 68.1 (d, C(3')); 130.8, 137.0, 138.9 (3s, C(3), C(1'), C(2')); 139.4 (d, C(4)); 199.3 (s, C(2)). MS: 222 (5, M^+ ,

 $C_{14}H_{22}O_2), 207 (9), 204 (8), 189 (25), 161 (14), 137 (18), 123 (41), 105 (13), 91 (15), 77 (13), 55 (12), 43 (100), 41 (24). Anal. calc. for C_{14}H_{22}O_2 (222.33): C 75.63, H 9.97; found: C 75.49, H 9.91.$

(Z)-4-(3'-Hydroxy-2',6',6'-trimethyl-1'-cyclohexenyl)-3-methyl-3-buten-2-one ((Z)-13). ¹H-NMR (3:2 mixture of (Z)- and (E)-13): 1.03, 1.07 (2s, 2 CH₃-C(6')); 1.66 (m, J = 1.0, CH₃-C(2')); 1.97 (m, J = 1.5, CH₃-C(3)); 0.8-2.1 (m, 2H-C(4'), 2H-C(5'), OH); 2.21 (s, 3H-C(1)); 3.8-4.2 (m, H-C(3')); 6.32 (m, $w_{\frac{1}{12}} = 6$, H-C(4)).

(E)-4-(3'-Hydroxy-6',6'-dimethyl-2'-methylidenecyclohexylidene)-3-methyl-2-butanone, Isomer A (15A). B.p. 90°/0.01 Torr. UV (0.373 mg in 25 ml MeCN): 215 (sh, 5500). UV (4.450 mg in 10 ml MeCN): 290 (200). IR: 3620w, 3490w (br.), 3080w, 2960s, 2930s, 2870m, 2850m, 1710s, 1635w, 1470w (sh), 1455m, 1420m, 1380m, 1365m, 1355s, 1215m, 1170m, 1130w, 1070m, 1060m, 1045m, 1000w, 980w, 965m, 915s, 890w, 870w. ¹H-NMR: 1.02 (s, 2 CH₃-C(6')); 1.11 (d, J = 7, CH₃-C(3)); 1.00-2.10 (m, 2H-C(4'), 2H-C(5')); 2.11 (s, 3H-C(1)); 3.57 (qd, $J_1 = 7$, $J_2 = 10$, H-C(3)); 4.12 (m, $w_{Y_2} = 15$, H-C(3')); 4.74 (m, $w_{Y_3} = 5$, H-C=C(2')); 5.21 (d, J = 10, H-C(4)); 5.26 (m, $w_{Y_4} = 4$, H-C=C(2')). ¹³C-NMR (ca. 80% pure): 17.1, 26.2, 26.5, 27.3 (4q, 4 CH₃); 31.0, 36.7 (2t, C(4'); C(5')); 37.1 (s, C(6')); 46.2 (d, C(3)); 72.6 (d, C(3')); 109.7 (t, C=C(2')); 120.8 (d, C(4)); 147.7, 148.9 (2s, C(1'), C(2')); 211.1 (s, C(2)). MS: 222 (4, M^+ , C₁₄H₂₂O₂), 189 (11), 179 (42), 165 (12), 162 (26), 161 (80), 148 (22), 147 (66), 135 (21), 133 (23), 123 (58), 121 (24), 119 (54), 109 (22), 107 (35), 106 (100), 105 (62), 95 (27), 93 (23), 91 (34), 81 (28), 79 (20), 69 (28), 67 (17), 55 (37), 43 (72), 41 (35).

Isomer B (15B). B.p. 90°/0.01 Torr. UV (0.471 mg in 25 ml MeCN): 215 (4250). UV (1.311 mg in 2 ml MeCN): 292 (240). IR: 3620w, 3470w (br.), 3080w, 2960s, 2930s, 2870m, 2850m, 1710s, 1630w, 1460m (sh), 1450m, 1415w (sh), 1380w (sh), 1370w (sh), 1350m, 1210w, 1180w (sh), 1165m 1065m, 1050m (sh), 960w, 910m, 870w. ¹H-NMR: 1.00, 1.04 (2s, 2 CH₃-C(6')); 1.12 (d, J = 7.0, CH₃-C(3)); 1.0–2.1 (m, 2H–C(4'), 2H–C(5')); 2.04 (s, 3H–C(1)); 2.22 (m, $w_{1/2} = 10$, OH); 3.62 (qd, $J_1 = 7.0$, $J_2 = 10.0$, H–C(3)); 3.96–4.20 (m, H–C(3')); 4.76, 5.31 (2m with *t*-character, $w_{1/2} = 5$, 2H–C=C(2')); 5.18 (d, J = 10.0, H–C(4)). ¹³C-NMR: (ca. 90% pure): 17.2, 26.4, 27.5, 28.2 (4q, 4 CH₃); 31.9, 37.8 (2t, overlapping with *s* at 37.8, C(4'), C(5')); 37.8 (s, C(6')); 46.9 (d, C(3)); 73.2 (d, C(3')); 109.3 (t, C=C(2')); 121.3 (d, C(4)); 148.4, 150.0 (2s, C(1'), C(2')); 211.0 (s, C(2)). MS: 222 (4, M^+ , C₁₄H₂₂O₂, 207 (4), 204 (5), 189 (10), 179 (30), 161 (67), 148 (18), 147 (19), 133 (20), 123 (49), 119 (43), 107 (26), 106 (38), 105 (50), 95 (25), 93 (20), 91 (31), 81 (28), 79 (22), 77 (24), 69 (28), 55 (41), 43 (100), 41 (54). Anal. calc. for C₁₄H₂₂O₂ (222.33): C 75.63, H 9.97; found: C 75.74, H 10.05.

(E)-3-Hydroxy-6,6-dimethyl-2-methylidene-1-(1'-propenyl) cyclohexyl Methyl Ketone, Isomer A (16A). IR: 3610w, 3450w (br.), 3020w, 2960m (sh), 2950m, 2920s, 2870m, 1705s, 1635w, 1475w, 1450m, 1380m, 1360w, 1350m, 1270w, 1190w, 1170m, 1115s, 1040m, 985m, 940w, 910m, 900m (sh), 870w. ¹H-NMR: 0.73, 1.08 (2s, 2 CH₃-C(6)); 1.78 (dd, $J_1 = 6.0, J_2 = 1.5, 3H-C(3')$); 1.95 (s, CH₃CO), 0.80–2.40 (m, 2H–C(4), 2H–C(5)); 3.14 (m, $w_{1/2} = 10$, OH); 3.66–3.96 (m, H–C(3)); 5.41 (AB, $J = 16.0, \delta_A = 5.15, \delta_B = 5.73$, A part split to $q, J_1 = 6.0, H-C(2')$, B part split to $q, J_2 = 1.5, H-C(1')$); 5.26, 5.57 (2m, $w_{1/2} = 4.0, 2H-C=C(2)$). ¹³C-NMR: 18.6, 25.2, 25.7, 27.9 (4q, 4 CH₃); 32.4, 34.6 (2t, C(4), C(5)); 38.2 (s, C(6)); 67.1 (s, C(1)); 70.7 (d, C(3)); 109.8 (t, C=C(2)); 128.3, 128.8 (2d, C(1'), C(2')); 150.5 (s, C(2)); 109.3 (s, CH₃-O). MS: 222 (2, M^+ , C₁₁H₂₂O₂), 189 (11), 179 (19), 161 (22), 148 (11), 136 (26), 123 (19), 119 (29), 105 (22), 91 (18), 84 (24), 69 (17), 55 (24), 45 (28), 43 (100), 41 (37).

Isomer B (16B), (*ca.* 75% pure). IR: 3610*w*, 3460*w* (br.), 3080*w*, 3010*w* (sh), 2960*s*, 2925*s*, 2865*s*, 1705*s*, 1630*w*, 1440*w* (br.), 1380*m*, 1360*m*, 1345*m*, 1210*m*, 1170*m*, 1115*m*, 1045*m*, 980*m*, 965*m*, 910*s*. ¹H-NMR (80 MHz): 0.80, 1.15 (2*s*, 2 CH₃--C(6)); 1.80 (*dd*, $J_1 = 6.0$, $J_2 = 1.5$, 3H--C(3')); 2.06 (*s*, CH₃--CO); 0.70–2.50 (*m*, 2H--C(4), 2H--C(5), OH); 4.27-4.45 (*m*, H--C(3)); 5.46 (*AB*, J = 16.0, $\delta_A = 5.24$, $\delta_B = 5.68$, *A* part split to *q*, $J_1 = 6.0$, H--C(2'), *B* part split to *m*, $w_{ij} = 3$, H--C(1')); 5.20, 5.38 (2*m*, $w_{ij} = 3$, 2H--C=-C(2)). ¹³C-NMR: 18.7, 25.2, 25.5, 28.8 (4*q*, 4 CH₃); 30.6, 31.7 (2*t*, C(4), C(5)); 38.0 (*s*, C(6)); 72.9 (*d*, C(3)); 115.7 (*t*, *C*=C(2)); 129.3 (2*d*, C(1'), C(2')); 150.0 (*s*, C(2)); 208.5 (*s*, CH₃--CO). MS: 222 (5, M^+ , C₁₄H₂₂O₂), 204 (35), 189 (41), 161 (31), 159 (18), 148 (34), 147 (44), 136 (38), 135 (35), 133 (45), 119 (41), 105 (55), 91 (66), 85 (45), 83 (70), 79 (35), 77 (37), 55 (3), 47 (30), 43 (100), 41 (68).

4,5-Epoxy-1,4,8,8-tetramethylspiro[2.5]oct-1-yl Methyl Ketone (**35**). IR: 2960s, 2930s, 2870m, 1705s, 1450m, 1380m, 1355m. ¹H-NMR (300 MHz, CDCl₃): 0.73 (d, J = 5.5, H-C(2)); 0.87, 0.98 (2s, 2 CH₃-C(8)); 1.19 (s, CH₃-C(1)); 1.15-1.35 (m, H-C(7)); 1.51 (s, CH₃-C(4)); 1.55-1.75 (m, H-C(7)); 1.91 (d, J = 5.5, H-C(2)); 1.94-2.04 (m, H-C(6)); 2.04-2.16 (m, H-C(6)); 2.32 (s, CH₃CO); 2.87 (dm, J = 3.5, $w_{y_2} = 2.5$, H-C(5)). ¹³C-NMR (25 MHz, contains ca. 30% of (E)-2): 17.2 (dd, C(2)); 18.9, 24.4, 26.3, 27.1 (4q, 4 CH₃); 21.8, 31.9 (2t, C(7), C(6)); 34.5, 38.2, 42.1 (3s, C(1), C(3), C(8)); 59.5 (s, C(4)); 60.5 (d, C(5)); 207.0 (s, C=O).

2.5.2. At $\lambda = 254$ nm. A soln. of (E)-2 (180 mg, 0.81 mmol) in MeCN (15 ml) was irradiated (quartz, lamp A, 91% conversion). After chromatography (Et₂O/pentane 1:1), the following product distribution was determined (¹H-NMR) (Z)-2 (6%), 14A (2%), 14B (2%), (E)-13 (17%), (Z)-13 (30%), 15A (14%) and 15B (16%).

2.6. Photolyses of (E)-3 in MeCN. – 2.6.1. At $\lambda > 280$ nm. A soln. of (E)-3 (510 mg, 2.29 mmol) in MeCN (100 ml) was irradiated (Pyrex, lamp B, 93% conversion). Chromatography (Et₂O/pentane 1:1) gave mixed fractions from which the following product distribution was determined (¹H-NMR): (Z)-3 (39%), 18 (1%), (E)-17 (16%), (Z)-17A (13%), (Z)-17B (25%), and additional photoproducts arising from (E/Z)-17 (2%)⁷). Compounds (E)-17, (Z)-17A, and (Z)-17B were purified and isolated as their acetates.

(Z)-2,4,4-Trimethyl-3-(4'-oxo-2'-penten-2'-yl)-2-cyclohexenyl Acetate, Conformer A (Acetate of (Z)-17A). UV (0.601 mg in 25 ml MeCN): 233 (10600). UV (5.942 mg in 2 ml MeCN): 322 (75). IR: 3010w (sh), 2960m, 2940m, 2920m, 2870w, 2855w, 1730s, 1695m, 1665m, 1600m, 1470w, 1460w (sh), 1450m, 1435m, 1370s, 1360m (sh), 1240s, 1190m, 1170w, 1160w, 1135w, 1010m, 990w (sh), 960m, 940w, 875w. ¹H-NMR: 1.00, 1.11 (2s, 2 CH₃-C(4)); 1.55 (s, CH₃-C(2)); 1.98 (d, J = 1.5, 3H-C(1)); 2.08, 2.20 (2s, 3H-C(5'), CH₃CO₂); 0.80-2.30 (m, 2H-C(5), 2H-C(6)); 5.24 (m, $w_{1/2} = 10$, H-C(1)); 6.13 (m, $w_{1/2} = 4$, H-C(3')). ¹³C-NMR: 17.7, 21.2, 26.9, 27.6, 29.4, 29.8 (6q, 6 CH₃); 25.5, 35.4 (2t, C(5), C(6)); 34.7 (s, C(4)); 70.9 (d, C(1)); 124.6, 145.2, 151.9 (3s, C(2)), C(3), C(2')); 129.2 (d, C(3')); 170.9 (s, CH₃CO₂); 198.2 (s, C(4')); MS: 204 (27, $M^+ -$ CH₃CO₂H), 189 (22), 161 (36), 149 (32), *148* (100), 147 (25), 133 (74), 119 (24), 108 (31), 105 (50), 91 (29), 79 (17), 77 (21), 45 (29), 43 (87), 41 (28). Anal. calc. for C₁₆H₂₄O₃ (264.37): C 72.69, H 9.15; found: C 72.78, H 9.25.

Conformer B (Acetate of (Z)-17B). UV (0.604 mg in 25 ml MeCN): 233 (9900). UV (6.886 mg in 2 ml MeCN): 326 (70). IR: 3000w (sh), 2970w (sh), 2960m, 2940m, 2915w, 2860w, 1730s, 1690m, 1665w, 1600m, 1435w br., 1365m, 1360w (sh), 1350w, 1240s, 1195m, 1160w, 1010m, 955w, 935w. ¹H-NMR: 0.91, 1.13 (2s, 2 CH₃-C(4)); 1.50 (s, CH₃-C(2)); 1.98 (d, J = 1.5, 3H-C(1')); 2.08, 2.13 (2s, 3H-C(5'), CH₃CO₂); 0.80-2.30 (m, 2H-C(5), 2H-C(6)); 5.18-5.40 (m, H-C(1)); 6.16 (m, $w_{Y_2} = 4$, H-C(3')). ¹³C-NMR: 17.0, 21.3, 26.9, 27.0, 30.2, 30.5 (6q, 6 CH₃); 25.5, 36.2 (2t, C(5), C(6)); 35.0 (s, C(4)); 72.3 (d, C(1)); 124.9, 144.4, 152.7 (3s, C(2), C(3), C(2')); 127.6 (d, C(3')); 170.9 (s, CH₃CO₂), 197.4 (s, C(4')). MS: 204 (30, M^+ - CH₃CO₂H), 189 (22), 161 (34), 149 (31), 148 (100), 147 (22), 133 (68), 119 (21), 108 (27), 105 (45), 91 (30), 84 (22), 77 (23), 45 (30), 43 (87), 41 (33). Anal. calc. for C₁₆H₂₄O₃ (264.37): C 72.69, H 9.15; found: C 72.77, H 9.03.

(1' RS, 2' SR, 3' RS) - 4 - (2', 3' - Epoxy-2', 6', 6' - trimethylcyclohexyl) - 4-penten-2-one (18). (Obtained in a mixture containing 57% of (*E*)-3.) Signals assigned to 18: IR: 3080w, 1725m (sh), 1710s, 905m. ¹H-NMR: 0.86, 0.93 (2s, 2 CH₃-C(6')); 1.26 (s, CH₃-C(2')); 2.22 (s, 3H-C(1)); 0.80-2.30 (m, H-C(1'), 2H-C(4'), 2H-C(5')); 2.80-3.02 (m, H-C(3')); 3.29 (*AB*,*J* $= 15, <math>\delta_A$ = 3.08, δ_B = 3.50, 2H-C(3)); 5.05, 5.22 (2m, $w_{1/2}$ = 3, 2H-C(5)). ¹³C-NMR (75 MHz): 49.9 (*t*, C(3)); 54.3, 57.5 (2*d*, C(1'), C(3')); 58.3 (*s*, C(2')); 119.2 (*t*, C(5)); 140.7 (*s*, C(4)); 204.5 (*s*, C(2)).

2.6.2. At $\lambda = 254$ nm. A soln. of (E)-3 (629 mg, 2.83 mmol) in MeCN (150 ml) was irradiated (quartz, lamp A, 86% conversion). After chromatography (Et₂O/pentane 1:1 to 3:1), the following product distribution was determined (¹H-NMR): (Z)-3 (19%), 18 (8%), (E)-17 (12%), (Z)-17A (22%), (Z)-17B (15%), and additional photoproducts arising from (E/Z)-17 (4%)⁷).

3. Additional Experiments. - 3.1. *Photooxygenation of* **10A** *and* **10B**. *a*) A suspension of **10A** (120 mg, 0.58 mmol), *Sensitox I* (*ca*. 20 mg) in CH₂Cl₂ (15 ml) was irradiated in the presence of O₂ (K_2 Cr₂O₇-filter, lamp *B*, conversion > 95%). Chromatography (Et₂O/pentane 2:1) yielded **20A** (65 mg, 47%). *b*) Analogously, photooxygenation of **10B** (106 mg, 0.51 mmol) afforded **20B** (70 mg, 57%).

5-Hydroxy-2,2,6,8-tetramethyl-7,9-10-trioxatricyclo[6.2.2.0^{1,6}]dodec-11-ene, Isomer A (**20A**). IR: 3575m, 3060w, 3010m (sh), 2980m (sh), 2960s (sh), 2940s, 2870m, 1740w (sh), 1720m (sh), 1700m, 1675s, 1655m (sh), 1615w (br.), 1465m (sh), 1445m, 1390m (sh), 1380s, 1375s (sh), 1365m, 1350m (sh), 1330m, 1310w, 1295m, 1245m, 1195m, 1170m, 1160s, 1105s, 1070s, 1030m, 1005m, 970s, 935m, 930m, 920w (sh), 900w, 870m, 850m. ¹H-NMR (80 MHz): 1.03, 1.15, 1.15, (3s, 2 CH₃-C(2), CH₃-C(6)); 1.58 (s, CH₃-C(8)); 1.00-2.40 (m, 2H-C(3), 2H-C(4)); 3.45-3.65 (m, OH); 3.70-3.90 (m, H-C(5)); 6.53 (*AB*, *J* = 8.0, δ_A = 6.45, δ_B = 6.60, H-C(11), H-C(12)). ¹³C-NMR: 19.6, 20.8, 24.5, 26.1 (4q, 4 CH₃); 27.0, 33.7 (2t, C(3), C(4)); 35.6 (s, C(2)); 73.5 (d, C(5)); 81.2, 83.2, 95.0 (3s, C(1), C(6), C(8)); 128.7, 133.4 (2d, C(11), C(12)).

Isomer B (20B). IR: 3610*m*, 3480*w* (br.), 3060*w*, 2990*m*, 2950*s*, 2930*s* (sh), 2870*m*, 1740*s*, 1735*s* (sh), 1725*m* (sh), 1715*m* (sh), 1625*w* (br.), 1475*w* (sh), 1465*m* (sh), 1460*m*, 1445*m*, 1380*s*, 1350*m* (sh), 1325*m*, 1285*w*, 1260*m*, 1240*s*, 1205*m*, 1195*m*, 1170*m*, 1155*m*, 1115*s*, 1100*s*, 1075*s*, 1040*s*, 1015*s*, 980*m*, 965*s*, 940*w*, 925*w*, 905*s*, 875*m*, 855*w*. ¹H-NMR (80 MHz): 1.00, 1.08, 1.12 (3*s*, 2 CH₃–C(2), CH₃–C(6)); 1.52 (*s*, CH₃–C(8)); 1.00–2.50 (*m*, 2H–C(3), 2H–C(4), OH); 4.20–4.50 (*m*, H–C(5)); 6.47 (*AB*, *J* = 8.0, δ_{A} = 6.40, δ_{B} = 6.54, H–C(11), H–C(12)). ¹³C-NMR (70% pure): 19.5, 20.7, 24.4, 26.1 (4*q*, 4 CH₃); 27.0, 33.7 (2*t*, C(3), C(4)); 35.5 (*s*, C(2)); 73.4 (*d*, C(5)); 81.0, 83.1, 94.9 (3*s*, C(1), C(6), C(8)); 128.6, 133.4 (2*d*, C(11), C(12)).

3.2. Acid-Catalyzed Transformation of 20A and 20B to 24. A soln. of 20A (107 mg, 0.45 mmol) in THF (15 ml) and aq. HCl (1M, 1.5 ml) was stirred for 20 h at r.t. Chromatography (AcOEt/CH₂Cl₂/pentane 2:1:1)

yielded 24 (26 mg, 24%). The analogous transformation of 20B (86 mg, 0.36 mmol) in THF (15 ml) and aq. HCl (1M, 1.5 ml) afforded 24 (19 mg, 22%), which was purified as its acetate.

1-Acetyl-4,4-dimethyl-5,8-dioxo-6-nonenyl Acetate (Acetate of **24**). B.p. 110°/0.01 Torr. UV (0.233 mg in 25 ml MeCN): 233 (12100). UV (2.150 mg in 2 ml MeCN): 342 (80), end absorption to 415. IR: 3010w (sh), 2965*m*, 2930*m*, 2870*w*, 1745*s*, 1735*s*, 1705*s* (sh), 1685*s*, 1615*w*, 1470*m*, 1425*m*, 1390*m*, 1370*s*, 1360*s*, 1320*m*, 1285*s*, 1260*m* (sh), 1230*s*, 1190*m*, 1160*m*, 1155*m* (sh), 1085*m* (sh), 1065*m*, 1040*m*, 980*m*. ¹H-NMR: 1.16 (*s*, 2 CH₃-C(4)); 1.63 (*m*, $w_{Y_4} = 6$, 2H-C(3), 2H-C(2)); 2.11, 2.11, 2.34 (3*s*, 3H-C(9), CH₃-CO, CH₃CO₂); 4.86–5.00 (*m*, H-C(1)); 7.11 (*AB*, *J* = 15, $\delta_A = 6.96$, $\delta_B = 7.26$, H-C(6), H-C(7)). ¹³C-NMR: 20.6, 23.7, 23.7, 26.1, 29.3 (5*q*, 5 CH₃); 25.5, 34.1 (2*t*, C(2), C(3)); 46.7 (*s*, C(4)); 78.4 (*d*, C(1)); 132.2, 137.6 (2*d*, C(6), C(7)); 170.5 (*s*, CH₃CO₂); 197.7, 203.5, 204.8 (3*s*, C(5), C(8), CH₃CO). MS: 240 (2, *M* ⁺ - C₂H₂O), 239 (3), 197 (7), 125 (23), 98 (43), 43 (100). Anal. calc. for C₁₅H₂₂O₅ (282.34): C 63.81, H 7.85; found: C 63.67, H 7.97.

3.3. Catalytic Hydrogenation of **25B**. A soln. of **25B** (30 mg, 0.11 mmol) in EtOH (1 ml) was added to an activated (H_2 -saturated) suspension of Pd/BaSO₄ (*ca.* 5 mg) in EtOH (3 ml) and stirred for 2 h at ambient temp. The catalyst was filtered off and the solvent removed *in vacuo* to give a quantitative yield of the pure product **26** [13].

3.4. Photooxygenation of 14A and 14B. Solns. of 14A (93 mg, 0.42 mmol) and 14B (75 mg, 0.34 mmol) in CH_2Cl_2 (15 ml) were photooxygenated as described in *Sect. 3.1* affording 21A (50 mg, 47%), and 21B (45 mg, 52%), respectively.

5-Hydroxy-2,2,6,8,12-pentamethyl-7,9-10-trioxatricyclo[6.2.2.0^{1.6}]dodec-11-ene, Isomer A (**21A**). IR: 3580m, 3050w, 3000m (sh), 3060s (sh), 3045s, 3025s (sh), 2890m (sh), 2870m, 1750m (br.), 1475w (sh), 1465m (sh), 1450s, 1435m, 1395s, 1385s, 1375s, 1335w, 1290w, 1340m, 1215m, 1185m, 1165m, 1125s, 1085s, 1075s, 1025m, 970s, 930m. ¹H-NMR: 1.00, 1.08, 1.10 (3s, 2 CH₃-C(2), CH₃-C(6)); 1.50 (s, CH₃-C(8)); 1.93 (d, J = 1.5, CH₃-C(12)); 0.90-1.20 (m, 2H-C(3), 2H-C(4)); 3.50-3.90 (m, H-C(5), OH); 6.20 (m, $w_{Y_2} = 5$, H-C(11)). ¹³C-NMR: 16.8, 18.2, 24.4, 26.3, 26.3 (5q, 5 CH₃); 26.0, 29.1 (2t, C(3), C(4)); 35.4 (s, C(2)); 71.4 (d, C(5)); 78.2, 83.7 (2s, C(1), C(6)); 97.5 (s, C(8)); 123.0 (d, C(11)); 141.2 (s, C(12)). MS: 254 (6, M^+ , C₁₄H₂₂O₄), 236 (2), 211 (4), 193 (16), 179 (6), 167 (8), 151 (14), 139 (23), 137 (21), 125 (14), 121 (12), 111 (22), 109 (22), 95 (15), 81 (15), 71 (15), 69 (18), 55 (19), 43 (100), 41 (28).

Isomer B (21B). IR: 3610*m*, 3490*w* (br.), 3050*w*, 3000*m*, 2970*s* (sh), 2950*s*, 2930*m*, 2900*m* (sh), 2870*m*, 1655*w*, 1475*w* (sh), 1470*m* (sh), 1460*m* (sh), 1450*m*, 1440*m*, 1395*m*, 1385*s*, 1375*s*, 1325*m*, 1275*w*, 1340*w*, 1220*w*, 1180*w*, 1145*m*, 1120*s*, 1100*s*, 1075*m*, 1045*s*, 1015*m*, 960*s*, 910*s*, 875*m*, 850*w*. ¹H-NMR: 0.97, 1.02, 1.04 (3*s*, 2 CH₃-C(2), CH₃-C(6)); 1.43 (*s*, CH₃-C(8)); 1.90 (*d*, *J* = 1.5, CH₃-C(12)); 0.90–2.20 (*m*, 2H-C(3), 2H-C(4), OH); 4.22–4.46 (*m*, H-C(5)); 6.14 (*m*, $w_{1/2}$ = 5, H-C(11)). ¹³C-NMR: 16.7, 18.3, 19.6, 24.5, 26.1 (5*q*, 5 CH₃); 27.0, 33.7 (2*t*, C(3), C(4)); 35.6 (*s*, C(2)); 73.4 (*d*, C(5)); 80.9, 83.7 (2*s*, C(1), C(6)); 97.3 (*s*, C(8)); 122.6 (*d*, C(11)); 140.8 (*s*, C(12)). MS: 254 (13, *M* ⁺, C₁₄H₂₂O₄), 193 (13), 167 (8), 151 (16), 139 (31), 137 (22), 125 (10), 123 (13), 121 (16), 111 (27), 109 (27), 95 (18), 84 (28), 69 (20), 55 (20), *43* (100), 41 (28).

3.5. Equilibration of the Conformers (Z)-17A and (Z)-17B. Two samples of (Z)-17A and (Z)-17B (50 mg, 0.19 mmol) in CD₃CN (0.5 ml) were heated in an oil bath to 80° and the reaction was followed by ¹H-NMR. After 5 h both samples contained a *ca*. 1:1 mixture of (Z)-17A and (Z)-17B as the only products.

3.6. Treatment of 18 with Base. A mixture of (E)-3 and 18 (ca. 1:1, 230 mg, 1.03 mmol), NaOMe (230 mg, 4.26 mmol) and MeOH (10 ml) was heated under reflux for 5 h. The mixture was concentrated *in vacuo* and the residue extracted with Et₂O/pentane 1:1. Workup afforded pure (E)- and (Z)-3 (82:18, 215 mg, 93%).

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